



**SVKM International School, Mumbai**  
**Long Term Plan for CIE AS Level**  
**SCHEME OF WORK**  
**Batch: 2019-21**

**Teacher: Mr. Manoj Kapoor**

**Subject: Chemistry**

Week and Month	Topic	Sub topics & Learning Outcomes	Teaching activities / Integration of ICT components	Assessment Summative / formative	Course work / practical component	Resources
<b>JULY</b> <b>(3 Week)</b>	<b>1. Atoms, molecules and stoichiometry</b>  <b>2. Atomic structure</b>	1.2 The mole and the Avogadro constant 1.2 The mole and the Avogadro constant 1.3 The determination of relative atomic masses, Ar 1.4 The calculation of empirical and molecular formulae 1.5 Reacting masses and volumes (of solutions and gases)  2.1 Particles in the atom 2.2 The nucleus of the atom 2.3 Electrons: energy levels, atomic orbitals, ionisation energy, electron affinity	Use of Ppts. Scientific American Research Paper related to Avogadro's No and it's Discovery.	1. Worksheets  2. Class Tests after each of the topics are completed.  3. Solving topics related questions from past papers	<b>Textbooks</b> Lainchbury experiment 1.1  <b>Practical Booklets</b> 2, 3  <b>Textbooks</b> Lainchbury, Experiment 1.3 Hill Practical 1 & 4 Hutchings experiment 52  <b>Online</b> <a href="http://www.rsc.org/learn-chemistry/resource/res00000697/titratin">http://www.rsc.org/learn-chemistry/resource/res00000697/titratin</a>	<a href="http://www.rsc.org/learn-chemistry/resource/res00000401/the-determination-of-relative-atomic-mass">http://www.rsc.org/learn-chemistry/resource/res00000401/the-determination-of-relative-atomic-mass</a>  Online <a href="http://www.s-cool.co.uk/a-level/chemistry/(atomic%20structure)">http://www.s-cool.co.uk/a-level/chemistry (atomic structure)</a> <a href="http://www.rsc.org/learn-chemistry/resource/res00001100/an-analogy-for-the-atom">www.rsc.org/learn-chemistry/resource/res00001100/an-analogy-for-the-atom</a> Online <a href="http://www.youtube.com/watch?v=m92QR7CBNoQ">www.youtube.com/watch?v=m92QR7CBNoQ</a> <a href="http://www.youtube.com/watch?v=PpOAlj7sOEc">www.youtube.com/watch?v=PpOAlj7sOEc</a> Online



					<p><u>g-sodium-hydroxide-with-hydrochloric-acid</u>  <a href="http://www.rsc.org/learn-chemistry/resource/res00000436/to-find-the-formula-of-hydrated-copper-ii-sulfate">http://www.rsc.org/learn-chemistry/resource/res00000436/to-find-the-formula-of-hydrated-copper-ii-sulfate</a></p>	<p>www.chemguide.co.uk (atoms)            Past papers            Paper 11, June 2013, Q5            Paper 13, June 2013, Q2            Paper 12, June 2012, Q2            Paper 13, Nov 2013, Q31            Paper 23, Nov 2013, Q3 (b)(i)            Online            www.chemguide.co.uk (atoms)            www.youtube.com/watch?v=2AFPfg0Como  <a href="http://www.kentchemistry.com/links/AtomicStructure/PauliHundsRule.htm">http://www.kentchemistry.com/links/AtomicStructure/PauliHundsRule.htm</a></p> <p>Past papers            Paper 11, Nov 2011, Q3            Paper 11, Nov 2012, Q2            Paper 11, June 2012, Q1            Paper 12, June 2012, Q4            Paper 12, June 2013, Q31            Paper 13, June 2012, Q2            Paper 13, June 2013, Q32</p>
<b>AUGUST (4 Weeks)</b>	<p><b>3. Chemical bonding</b></p> <p><b>4. States of matter</b></p>	<p>3.1 Ionic bonding            3.2 Covalent bonding and coordinate (dative covalent) bonding including shapes of simple molecules            3.3 Intermolecular forces, electronegativity and bond properties            3.4 Metallic bonding            3.5 Bonding and physical properties</p> <p>4.1 The gaseous state: ideal and</p>	<p>Use of Ppts for VSEPR            Videos related to Chemical Bondings and its applications.</p>	<p>1. Worksheets            2. Class Tests after each of the topics are completed.            3. Solving topics related questions from past papers</p>	<p>Textbooks            Lainchbury            Experiments 3.4 &amp; 3.5</p>	<p>Past Papers            Paper 21, Nov 2013, Q1 (b)            Paper 23, Nov 2013, Q1 (a) &amp; Q3 (b)(ii)            Paper 41, Nov 2013, Q1 (a)            Paper 43, Nov 2013, Q1 (a)</p> <p>Past Papers            Paper 11, June 2013, Q33            Paper 21, Nov 2013, Q1 (a)</p>



	<b>5. Chemical energetics</b>	<p>real gases and <math>pV = nRT</math></p> <p>4.2 The liquid state</p> <p>4.3 The solid state: lattice structures</p> <p>5.1 Enthalpy change, <math>\Delta H</math></p> <p>5.2 Hess' Law, including Born-Haber cycles</p>				<p>Textbooks</p> <p>Lainchbury Experiment 3.6</p> <p>Hutchings Experiment 26</p> <p>Online</p> <p><a href="http://www.rsc.org/learn-chemistry/resource/res00000421/particles-in-motion">http://www.rsc.org/learn-chemistry/resource/res00000421/particles-in-motion</a></p> <p><a href="http://www.rsc.org/learn-chemistry/resource/res00000390/the-energetics-of-freezing">http://www.rsc.org/learn-chemistry/resource/res00000390/the-energetics-of-freezing</a></p>
<b>SEPTEMBER (4 Weeks)</b>	<b>6. Electrochemistry</b>	<p>6.1 Redox processes: electron transfer and changes in oxidation number (oxidation state)</p> <p>7.1 Chemical equilibria: reversible reactions, dynamic equilibrium</p>	Use of Ppts for Electrochemistry. Videos related to Electrolysis and its applications.	<p>1. Worksheets</p> <p>2. Class Tests after each of the topics are completed.</p> <p>3. Solving topics related questions from past papers</p>	<p><b>Textbooks</b></p> <p>Lainchbury, Experiment 3.3 Hill S4</p> <p>Practical Booklet 4</p> <p>Textbooks Lainchbury Experiments 2.1, &amp; 2.2</p> <p>Textbooks Lainchbury Experiment 9.7</p> <p>Online <a href="http://www.chemguide.co.uk">www.chemguide.co.uk</a></p>	<p>Past Paper Paper 43, Nov 2013, Q2 (a)</p> <p>Past Paper Paper 43, Nov 2013, Q2 (c)(i)</p> <p>For Redox Practical Booklet 4, 5</p> <p>Past Papers Paper 11, June 2013, Q1 Paper 13, Nov 2013, Q1</p> <p>For Equilibrium:</p> <p>Online <a href="http://www.rsc.org/learn-chemistry/resource/res00000117/afl-equilibrium-reactions">http://www.rsc.org/learn-chemistry/resource/res00000117/afl-equilibrium-reactions</a></p> <p><a href="http://www.rsc.org/learn-chemistry/resource/res00000117/afl-equilibrium-reactions">http://www.rsc.org/learn-</a></p>
	<b>7. Equilibria</b>	<p>8.1 Simple rate equations,</p> <p>8.2 Effect of temperature: on reaction rates and rate constants and the concept of activation energy</p>				
	<b>8. Reaction kinetics</b>	8.3 Homogeneous and heterogeneous catalysts including enzymes				



					<p>Textbooks Lainchbury Experiment 6.1</p> <p>Past Papers Paper 13, Nov 2013, Q34 Paper 21, June 2013, Q2 (b)</p>	<p>chemistry/resource/res00000001 /cobalt-equilibrium</p> <p>Past Paper Paper 41, Nov 2013, Q3 (b)</p> <p>For Kinetics: Practical Booklet 5</p> <p>Past Papers Paper 11, June 2013, Q8 Paper 13, Nov 2013, Q5 Past Paper Paper 12, June 2013, Q5 Many examples of questions involving the Boltzmann distribution can be found on past paper variants 1, 11, 12 and 13.</p>
<b>OCTOBER (3 Weeks)</b>	<p><b>9. The Periodic Table chemical periodicity</b></p> <p><b>10. Group 2</b></p>	<p>9.1 Periodicity of physical properties of the elements in Period 3 9.2 Periodicity of chemical properties of the elements in Period 3 9.3 Chemical periodicity of other elements</p> <p>10.1 Similarities and trends in the properties of the Group 2 metals, magnesium to barium, and their compounds 10.2 Some uses of Group 2 compound</p> <p>11.1 Physical properties of the</p>	<p>Use of Ppts. Videos related to alkali metals and its reactivity .</p>	<p>1. Worksheets</p> <p>2. Class Tests after each of the topics are completed.</p> <p>3. Solving topics related questions from past papers</p>		



	<b>11. Group 17</b>	Group 17 elements 11.2 The chemical properties of the elements and their hydrides 11.3 Some reactions of the halide ions 11.4 The reactions of chlorine with aqueous sodium hydroxide  11.5 Some important uses of the halogens and of halogen compounds				
<b>NOVEMBER ( 3 Weeks)</b>	<b>13. Nitrogen and sulphur</b>  <b>14. An introduction to organic chemistry</b>	13.1 Nitrogen ( Ammonia) 13.2 Sulfur: the formation of atmospheric sulfur dioxide, its role in acid rain  14.1 Formulae, functional groups and the naming of organic compounds 14.2 Characteristic organic reactions 14.3 Shapes of organic molecules; $\sigma$ and $\pi$ bonds 14.4 Isomerism: structural and stereoisomerism	Use of Ppts. Videos related to Haber's process and its applications.	1. Worksheets  2. Class Tests after each of the topics are completed.  3. Solving topics related questions from past papers		<a href="http://www.chemguide.co.uk/basics/isomerism/structural.html">http://www.chemguide.co.uk/basics/isomerism/structural.html</a> Past Paper Paper 23, Nov 2013, Q4 (a)(i) Past Papers Paper 11, June 2013, Q20 Paper 13, Nov 2013, Q21 Paper 43, Nov 2013, Q6 (f) Online <a href="http://www.rsc.org/learn-chemistry/resource/res00001261/khan-academy-stereochemistry">http://www.rsc.org/learn-chemistry/resource/res00001261/khan-academy-stereochemistry</a>
<b>DECEMBER (4 Weeks)</b>	<b>15. Hydrocarbons</b>  <b>16. Halogen derivatives</b>	15.1 Alkanes 15.2 Alkenes 15.3 Hydrocarbons as fuels  16.1 Halogenoalkanes 16.2 Relative strength of the C-Hal bond	Use of Ppts. Videos related to Fractional Distillations Cracking and its applications.	1. Worksheets  2. Class Tests after each of the topics are completed.  3. Solving topics related	Textbooks Hill Practical 28 Lainchbury Experiments 5.3 & 5.4	Online <a href="http://www.rsc.org/learn-chemistry/resource/res00000028/polythene">http://www.rsc.org/learn-chemistry/resource/res00000028/polythene</a> Online <a href="http://www.rsc.org/learn-chemistry/resource/res00000283/spectroscopy-in-a-suitcase-ir-student-resources">http://www.rsc.org/learn-chemistry/resource/res00000283/spectroscopy-in-a-suitcase-ir-student-resources</a>



				questions from past papers		Past Paper Paper 11, June 2013, Q23 Online <a href="http://www.rsc.org/learn-chemistry/resource/res00000115/afl-nucleophilic-substitution-reaction-mechanisms">http://www.rsc.org/learn-chemistry/resource/res00000115/afl-nucleophilic-substitution-reaction-mechanisms</a> Online <a href="http://www.rsc.org/learn-chemistry/resource/res00000454/making-a-photographic-print">http://www.rsc.org/learn-chemistry/resource/res00000454/making-a-photographic-print</a>
<b>JANUARY (4 Weeks)</b>	<b>17. Hydroxy compounds</b>  <b>18. Carbonyl compounds</b>	17.1 Alcohols 17.2 Phenol  18.1 Aldehydes and ketones	Use of Ppts. Videos related to Alcohol manufacturing	1. Worksheets  2. Class Tests after each of the topics are completed.  3. Solving topics related questions from past papers		Practical Booklet 6  Textbooks Hill Practical 30 Lainchbury Experiment 8.1  Online <a href="http://www.rsc.org/learn-chemistry/resource/res00000463/properties-of-alcohols">http://www.rsc.org/learn-chemistry/resource/res00000463/properties-of-alcohols</a> <a href="http://www.rsc.org/learn-chemistry/resource/res00000553/the-oxidation-of-alcohols">http://www.rsc.org/learn-chemistry/resource/res00000553/the-oxidation-of-alcohols</a>  Past Papers Paper 13, Nov 2013, Q27 Paper 23, Nov 2013, Q5 (c)  Online <a href="http://www.rsc.org/learn-chemistry/resource/res00000588/three-isomeric-alcohols">http://www.rsc.org/learn-chemistry/resource/res00000588/three-isomeric-alcohols</a>  Past Paper



						Paper 23, Nov 2013, Q5 (b)
<b>FEBRUARY (4 Weeks)</b>	<b>19. Carboxylic acids and derivatives</b>	19.1 Carboxylic acids 19.2 Acyl chlorides* 19.3 Esters	Use of Ppts. Videos related to Esters and its application in food Industries.	1. Worksheets 2. Class Tests after each of the topics are completed. 3. Solving topics related questions from past papers		
<b>MARCH (4 Weeks)</b>	<b>Revision</b>					
<b>APRIL</b>	<b>Mock Exams</b>					
<b>MAY</b>	<b>Board Exams</b>					